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A polymer is a substance composed of giant molecules that have been formed by the union of a large number of single molecules with one another. The word polymer is derived from the greek word "poly", meaning many, and "meros" meaning parts.

A polymer is composed of many monomer units joined together via a chemical reaction called polymerization. The degree of polymerization is the number of monomer units in a polymer molecule. It can be very long - this tremendous length gives polymers some of their unique properties.

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The molecular weight of a polymer is the degree of polymerization multiplied by the molecular weight of the monomer unit. Acrylonitrile monomer has a molecular weight of 71. A polymer molecule with a degree of polymerization of 100,000 is made up of 100,000 acrylonitrile monomer units and will have a molecular weight of 7.1 million ($100,000 \times 71$).

Organic polymers (or polyelectrolytes) are essentially water soluble linear polymers with molecular weights as low as a few hundred thousand and as high as ten million or greater. These products are characterized by the existence of ionized (electrically charged) site groups on the polymer molecule. The electrical charge on the polymer molecule can be negative (anionic), positive (cationic) or no charge (nonionic).

Polymer Types:

DIY POLYMER

APPEARANCE:

- Granular
- Bead

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